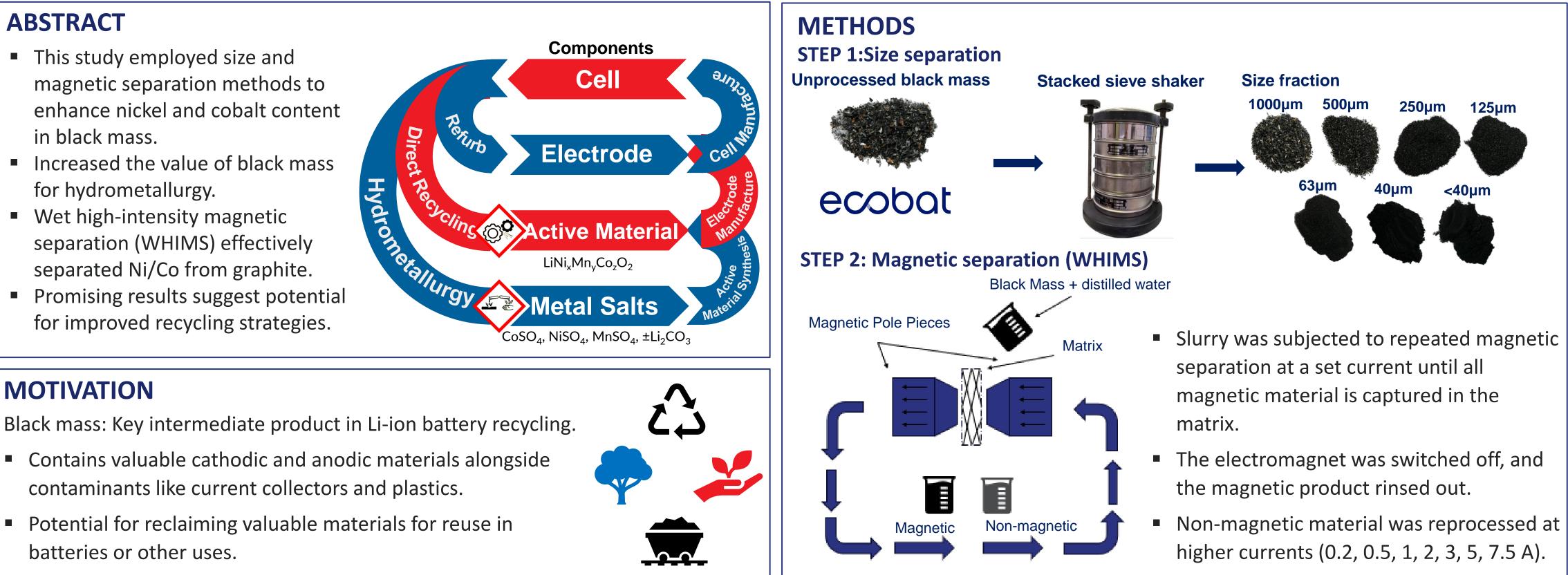
Banking on Better Black Mass

Investigating the use of different physical separation techniques to increase the value of black mass from spent Li-ion batteries

Kaja Glazer^{1,3}, Dr Rob Sommerville^{2,3}, Dr Elizabeth H. Driscoll^{2,3}, Prof Emma Kendrick^{2,3}

¹School of Chemistry, University of Birmingham, B15 2TT ²School of Metallurgy and Materials, University of Birmingham, B15 2TT ³The Faraday Institution, OX11 ORA



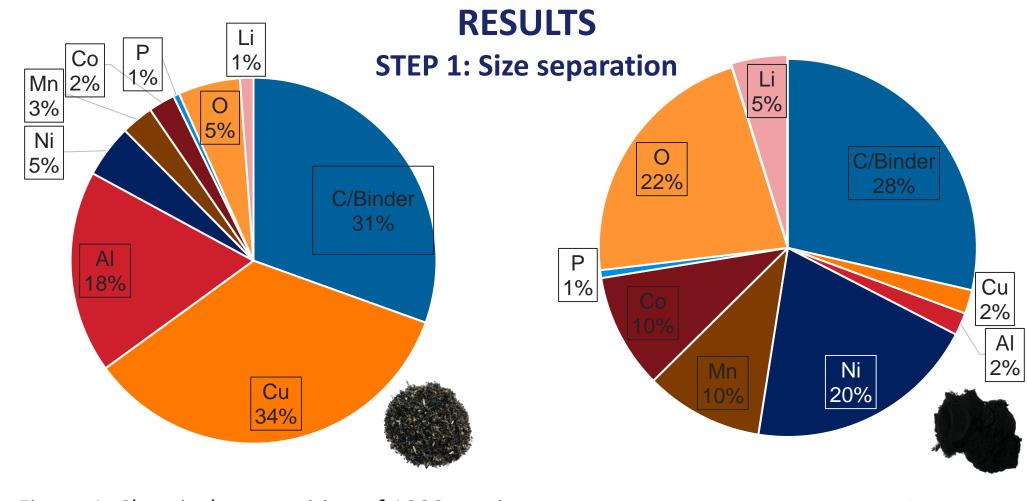


Figure 1. Chemical composition of 1000µm size Figure 2. Chemical composition of <40µm size fraction black mass.

fraction black mass.

Key findings:

- 30% reduction in contaminants (Cu, Al) from coarse to fine size.
- Elevated Valuable/ critical materials (Ni, Co, Graphite) in finer sizes. Why minimise contamination?
- Reducing Al and Cu levels improves the cost-effectiveness of hydrometallurgical recycling.
- By separating Al & Cu they can be recycled effectively, rather than discarded as waste during hydrometallurgical recycling.²

STEP 2: Magnetic separation

Component Recovery Grade

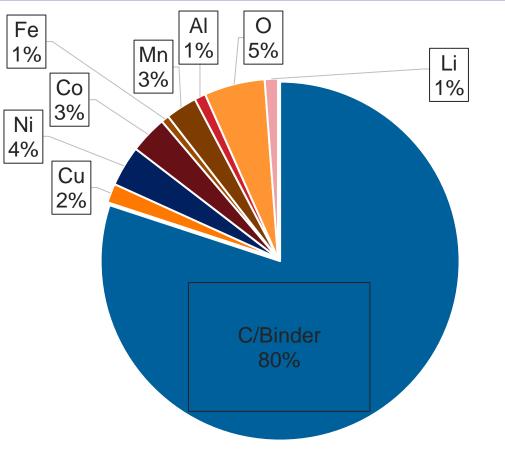


Figure 3. Chemical composition of the nonmagnetic product after WHIMS of <40µm sieved black mass.

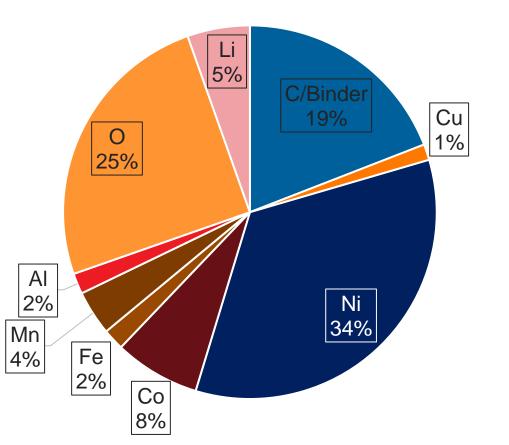


Figure 4. Chemical composition of the magnetic product at 0.2 A after WHIMS of <40µm sieved black mass. Graphite

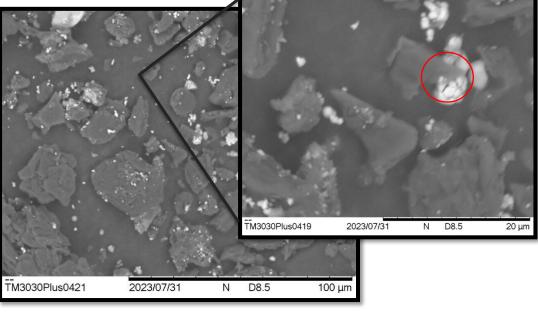


Figure 5: SEM of Non-magnetic fraction from WHIMS used sieved <40µm black mass.

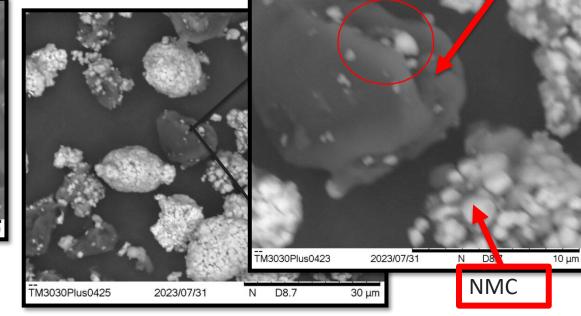


Figure 6. SEM of Magnetic concentrate separated at 0.2 A in WHIMS- sieved <40µm black mass.

Key findings:

- The separation was performed on the finest fraction.
- Effective separation of graphite and NM was achieved by exploiting their differin magnetic susceptibilities.³
- Grade of Ni was increased from 20 to 34% respectively. Graphite from 28 to 80% (See Table 1).

		(Wt%)	(Wt%)
	Ni* in 0.2 A	50.4	34.0
/IC ng	Co* in 0.2A	26.5	8.0
	Graphite in Non- Magnetic	30.1	80.0
Table 1. Grade and recovery in wt % for selected			

fractions, metals in Wt %. * Metals are given as wt % but are in oxide form, which reduces the grade.

UNIVERSITY^{OF} BIRMINGHAM

Why is the magnetic separation not 100%?

- The attachment of NMC and graphite results in cross-contamination.
- During the recycling process, NMC particles became embedded in the graphite causing carry over.

CONCLUSIONS

- Increased the value of black mass by two physical separation techniques.
- Used simple sieving to separate Ni/Co and graphite from contaminants Cu/Al.
- Used a mineral processing technique in a novel field and applied it to battery recycling, resulting in cost-effective separation of Ni/Co from graphite.
- This pretreatment can help further recycling processes like hydrometallurgy be more cost and energy-efficient.

Impact / Next steps

- Address the NMC and graphite attachment by focusing on:
- Limiting intimate mixing of anode and cathode during shredding
- Ultrasonication to liberate NMC embedded in Graphite
- Selective leaching of NMC from Graphite

REUSE & RECYCLING OF LITHIUM ION

This helps to achieve the ultimate goal of direct recycling of Li-ion batteries.

References

[1] Sommerville, R. et al. (2020) 'A review of physical processes used in the safe recycling of lithium ion batteries', Sustainable Materials and Technologies

[2] Sommerville, R. et al. (2021) 'A qualitative assessment of lithium ion battery recycling processes', Resources, Conservation and Recycling, 165, p. 105219.

[3] Jorge (2019) Wet magnetic separation https://www.911metallurgist.com/wet-magneticseparation/#:~:text=The%20magnetic%20characteristics%20of% 20the%20particle%20affect%20the,a%20direct%20function%20o *f%20both%20H%20and%20dH%2Fdx*.

Kaja Glazer is studying MSci Chemistry at the University of Birmingham. Interested in battery recycling and wishes to undertake a PhD in a related field. Aspiring to become a researcher in the battery recycling industry.





*Kaja at the Ecobat Ltd site visit

